



# Laser-induced breakdown spectroscopy as a straightforward bioimaging tool for plant biologists; the case study for assessment of photon-upconversion nanoparticles in *Brassica oleracea* L. plant

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## ABSTRACT

The main purpose of this work is to thoroughly describe the implementation protocol of laser-induced breakdown spectroscopy (LIBS) method in the plant analysis. Numerous feasibility studies and recent progress in instrumentation and trends in chemical analysis make LIBS an established method in plant bioimaging. In this work, we present an easy and straightforward phytotoxicity case study with a focus on LIBS method. We intend to demonstrate in detail how to manipulate with plants after exposures and how to prepare them for analyses. Moreover, we aim to achieve 2D maps of spatial element distribution with a good resolution without any loss of sensitivity. The benefits of rapid, low-cost bioimaging are highlighted.

In this study, cabbage (*Brassica oleracea* L.) was treated with an aqueous dispersion of photon-upconversion nanoparticles (NaYF<sub>4</sub> doped with Yb<sup>3+</sup> and Tm<sup>3+</sup> coated with carboxylated silica shell) in a hydroponic short-term toxicity test. After a 72-hour plant exposure, several macroscopic toxicity end-points were monitored. The translocation of Y, Yb, and Tm across the whole plant was set by employing LIBS with a lateral resolution 100 μm. The LIBS maps of rare-earth elements in *B. oleracea* plant grown with 50 μg/mL nanoparticle-treated and ion-treated exposures showed the root as the main storage, while the transfer *via* stem into leaves was minimal. On the contrary, the LIBS maps of plants exposed to the 500 μg/mL nanoparticle-treated and ion-treated uncover slightly different trends, nanoparticles as well as ions were transferred through the stem into leaves. However, the main storage organ was a root as well.

## 1. Introduction

Many types of nanoparticles (NPs) have been increasingly used in commercial products as well as in many research areas. This leads not only to the NP-accumulation in the environment and within the food chain (Rodrigues et al., 2016) but also to unknown toxic effects at various organism/tissue/cellular levels (Magnuson et al., 2011). The evaluation of NPs toxicity, bioaccumulation, and translocation in diverse organisms is an extremely challenging task (Modlitbová et al., 2020a). This is caused by a great number of variables in physicochemical properties of NPs (composition, shape, size, surface charge, functional group and ligands, ...). Thus, it is impossible to establish robust standard protocols. Consequently, the real effects of NPs and the sites of

their bioaccumulation still remain unknown.

The precise localization of NPs through the plant tissues is of a paramount importance in order to reveal the relationship between the exact location of NPs and their toxic effect. The distribution of NPs and their bioimaging in plants is evaluated most often by using transmission electron microscopy (Wang et al., 2012), scanning electron microscopy (Vittori Antisari et al., 2015), scanning transmission electron microscopy (Schwabe et al., 2015), micro-X-ray fluorescence microscopy (Cui et al., 2014), or laser ablation inductively coupled plasma mass spectrometry (Ko et al., 2019). However, rarely has been laser-induced breakdown spectroscopy (LIBS) showed to be suitable for a spatial distribution of elements contained in NPs assessment within various plant organs in a few recent papers (Krajcarová et al., 2017; Modlitbová et al.,

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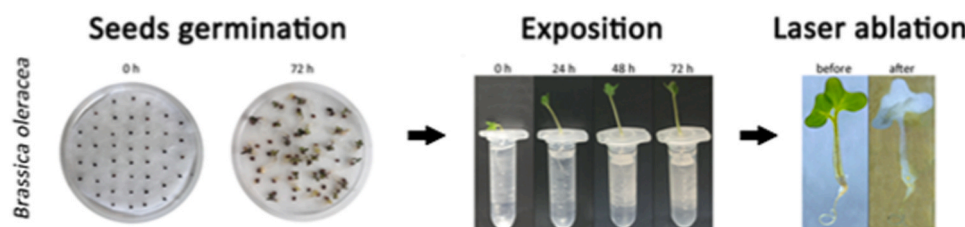


Fig. 1. Scheme of seeds germination, plant exposure, and plant samples before and after LIBS analysis.

2019, 2020b). Herein, LIBS represents an indispensable alternative to its analytical counterparts in the NPs bioimaging with a high resolution (micro-scale) on a large-scale sample (whole glass slide imaging).

The LIBS has been widely used in modern analytical chemistry. It is popular because of its advantages such as the fast turn-around time, multi-elemental capability (detection of light elements and halogens), analysis of samples in any state of matter, possibility of remote sensing, and the assessment of spatial elemental distribution on a large scale (so called imaging). Various biological samples were successfully analysed using LIBS method, e.g., mouse kidney, lung tumours, skin melanomas; summarized in two most recent reviews (Busser et al., 2018; Jolivet et al., 2019). LIBS was also applied for bioimaging of various NP types in plant tissues as mentioned above (Modlitbová et al., 2020a).

In this study, less common NPs, the rare-earth elements-doped photon-upconversion nanoparticles (UCNPs) are studied. The UCNPs are a new type of luminescent nanomaterial which started to be widely used as luminescent labels due to their unique optical properties (Zhou et al., 2015). They are composed of  $\text{NaYF}_4$  nanocrystals and doped with  $\text{Yb}^{3+}$  and  $\text{Tm}^{3+}$  ions; a carboxylated silica shell is added on the surface to increase the chemical stability and reduce the releasing of free ions (Modlitbová et al., 2019).

The cabbage (*Brassica oleracea* L.), which has been commonly used as a model crop plant in phytotoxicity and bioaccumulation experiments tests for a long time (Wang and Keturi, 1990), was employed during the whole study. This plant species belongs to ten recommended species by United States Environmental Protection Agency (US EPA) for the determination of toxicological effects of pesticides and other toxic substances (US EPA, 1996). Additionally, the cabbage belongs to the so-called hyperaccumulator plant family Brassicaceae (Reeves et al., 2018). Also, this plant was already used in the phytotoxicity assessment of various NPs, for example Ag NPs (Pokhrel and Dubey, 2013), ZnO NPs (Pokhrel and Dubey, 2013),  $\text{Al}_2\text{O}_3$  NPs (Amist et al., 2017), and CuO NPs (Singh et al., 2017). Moreover, the phytotoxicity of rare-earth elements containing NPs ( $\text{CeO}_2$ ,  $\text{La}_2\text{O}_3$ ,  $\text{Gd}_2\text{O}_3$ , and  $\text{Yb}_2\text{O}_3$  NPs) was established and the growth of cabbage root was negatively affected by  $\text{La}_2\text{O}_3$ ,  $\text{Gd}_2\text{O}_3$ , and  $\text{Yb}_2\text{O}_3$  NPs dispersions in a very high test concentration 2000  $\mu\text{g}$  NPs/mL (Ma et al., 2010).

In this study, the first objective was to evaluate the toxicity of the selected contaminants (UCNPs dispersion in two nominal concentrations 500 and 50  $\mu\text{g}$  UCNPs/mL, and two  $\text{Y}^{3+}$  and  $\text{Yb}^{3+}$  mixture solutions in corresponding concentrations serving as a positive control) to *B. oleracea* plant grow in an aqueous medium. After a 72-h exposure, we monitored four macroscopic end-points (root length, stem length, root + stem length, whole plant length). Then, the spatial distribution of Y, Yb, and Tm for UCNPs treated plants, and the spatial distribution of Y and Yb for positive control treated plants was set by LIBS method. The photon-upconversion laser microscanning was used as a complementary technique confirming the presence of Y, Yb, and Tm in the plants in the form of UCNPs (Sedlmeier et al., 2016).

The spatial distribution of Y, Yb (Tm) across the whole plant was established with a sufficient spatial resolution of 100  $\mu\text{m}$  by LIBS. The difference in behaviour between both contaminants as well as between both tested concentrations is clearly visible in obtained 2D maps. LIBS is a fast, straightforward, price sensitive (when compared to Laser Ablation Inductively Coupled Plasma Mass Spectrometry), and relatively

simple analytical method for semi-quantitative multi-element bioimaging in plants with a similar sensitivity to the photon-upconversion microscanner. Moreover, LIBS is a promising alternative to other techniques as it is able to detect NPs with no visible upconversion or fluorescence. It also detects NPs whose luminescence is quenched, e.g. because of changes in external conditions, such as pH values changes (Škarková et al., 2017).

## 2. Material and methods

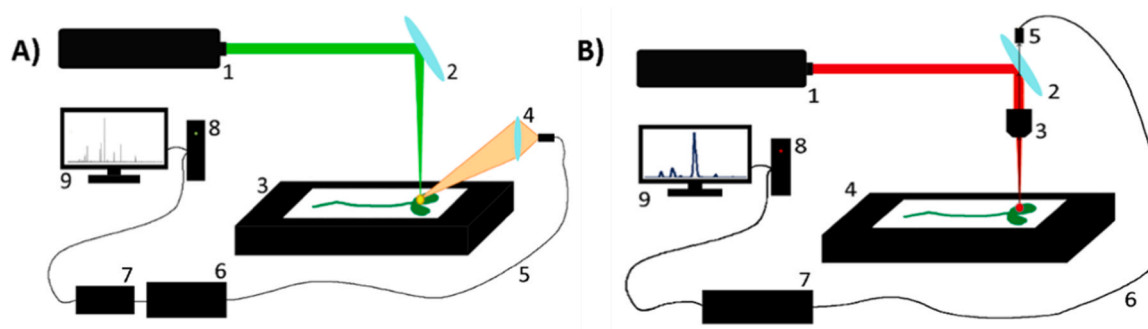
### 2.1. Synthesis and characterization of UCNPs

The preparation of UCNPs was described in detail in our previous work (Hlaváček et al., 2019), nothing was modified in this experiment. These were the used reagents:  $\text{YCl}_3 \times 6\text{H}_2\text{O}$  (99.99%),  $\text{YbCl}_3 \times 6\text{H}_2\text{O}$  (99.99%),  $\text{TmCl}_3 \times 6\text{H}_2\text{O}$  (99.99%), 1-octadecene (90%), oleic acid (90%), tetraethylorthosilicate (TEOS,  $\geq 99\%$ ), and Igepal CO-520 were purchased from Sigma-Aldrich (Steinheim, Germany) and used as starting materials without further purification. NaOH (p.a.), cyclohexane (p.a.), N,N-Dimethylformamide (p.a.), Methanol (p.a.), Acetone (p.a.), and  $\text{NH}_4\text{OH}$  25% (p.a.) were purchased from PENTA (Chrudim, Czech Republic). Carboxyethylsilanetriol, sodium salt; 25% (w/v) in water was obtained from ABCR GmbH (Karlsruhe, Germany). The nominal hydrodynamic particle diameter was determined with a Zetasizer Nano ZS (Malvern Instruments, Worcestershire, United Kingdom).

### 2.2. Plant exposure

Seeds of cabbage (*Brassica oleracea* L.) were purchased from MoravoSeed (Mikulov, Czech Republic). Firstly, cabbage seeds were germinated for 48 h in Petri dishes with the bottom covered with filtration paper in Milli-Q water (Millipore RG, Merck KGaA, Germany). Then seedlings with similar size (approximately 1 cm root length) were selected for the toxicity test. Plants were exposed for 72 h under the lighting cycle of 15 h light/9 h darkness at room temperature ( $22 \pm 1^\circ\text{C}$ ). Specially prepared Eppendorf tubes with a 2-mm hole on the top were filled with 2.0 mL of an exposure medium based on our previous studies (Modlitbová et al., 2020b). Each plant experiment consisted of a control group (Milli-Q water), two UCNPs dispersions in nominal concentrations 500  $\mu\text{g}$  UCNPs/mL (100  $\mu\text{g}$  Y/mL + 43.8  $\mu\text{g}$  Yb/mL) and 50  $\mu\text{g}$  UCNPs/mL (10  $\mu\text{g}$  Y/mL + 4.38  $\mu\text{g}$  Yb/mL), and two chloride mixture solutions at the same concentrations “500”  $\mu\text{g}$  /mL (100  $\mu\text{g}$  Y/mL + 43.8  $\mu\text{g}$  Yb/mL) and “50”  $\mu\text{g}$  /mL (10  $\mu\text{g}$  Y/mL + 4.38  $\mu\text{g}$  Yb/mL). Each exposure group contained 12 replicates of test plants. After the 72-hour exposure, the plants were photographed; plant roots, stems, and the overall length were measured. A simplified scheme of toxicity test followed by LIBS measurements is illustrated in Fig. 1.

The statistical analysis for root, stem, root + stem, and the whole plant length (5–11 plants per one treatment) was performed by OriginPro software (2015). The significance of the difference between the exposed and control plants was tested by the Mann-Whitney U-test. The level of significance was accepted at  $*p < 0.05$ ,  $**p < 0.01$ , and  $***p < 0.001$ .



**Fig. 2.** A) LIBS scheme: (1) laser, (2) focusing optics, (3) sample holder with plant sample, (4) collection optics, (5) optical fibre, (6) spectrometer, (7) detector, (8) PC, (9) example of LIBS emission spectra. 2B) Laser microscanner scheme: (1) laser and focusing optics, (2) dichroic mirror, (3) microscope objective, (4) sample holder with plant sample, (5) collection optics, (6) optical fiber, (7) spectrometer, (8) PC, (9) example of photon-upconversion spectra.

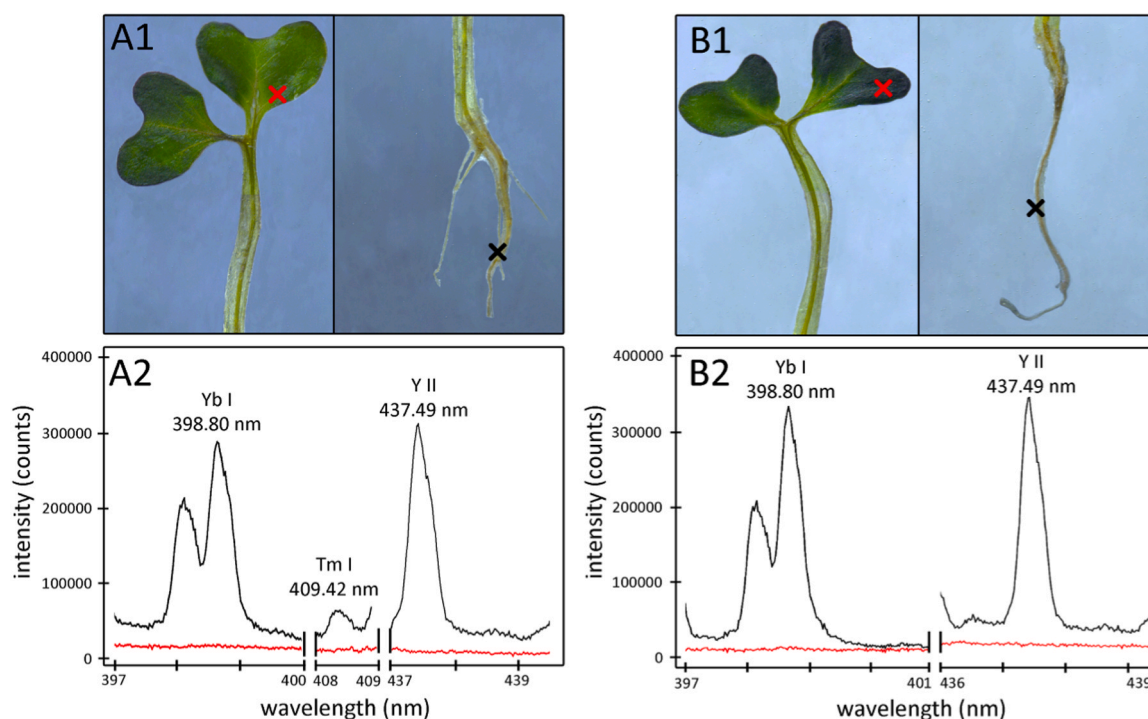
### 2.3. LIBS bioimaging

After the exposure, plants must be prepared for LIBS measurements (Jantzi et al., 2016). Several ways to prepare plants were already and summarized in an extensive review (Modlitbová et al., 2020a). These approaches are the most common. The first way is to use fresh plants and analyze them directly, even in *in-situ* conditions (Zhao et al., 2016). The main disadvantage is the water content in plants which is responsible for lower LIBS signals (Peng et al., 2017), and for no possibility of the experiment repetition. The second procedure is to prepare histological cross-sections from plant tissues (Krajcarová et al., 2017). This is a very time consuming procedure which requires a detailed optimization for various plant tissues, different contaminants as well as for different laser sources in LIBS set-ups. The third possibility is the most common, easy, and low-cost. It was used in this experiment. The plants were firstly thoroughly washed in Milli-Q water, then dried and molded at room

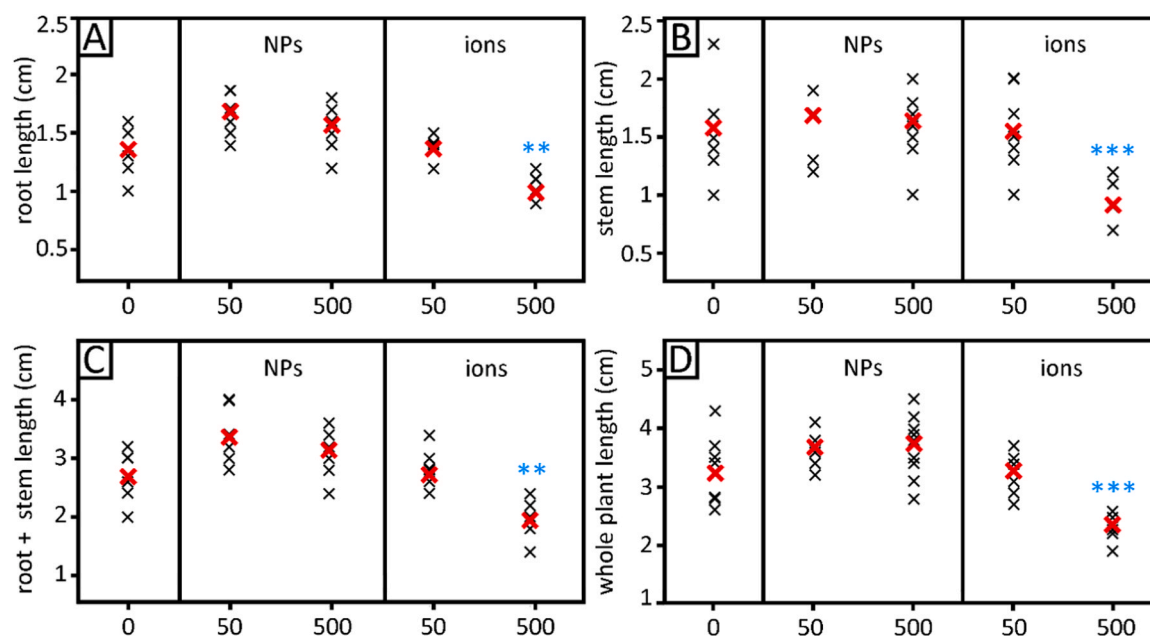
temperature for five days and then glued with epoxide onto the glass slide. Then, plants were photographed using an optical microscope (Stemi 2000-c, Zeiss, Oberkochen, Germany), then photon-upconversion microscans (only in case of UCNP treated plants), and LIBS measurements were performed.

The LIBS Discovery system (CEITEC, Brno, Czech Republic) was used for all experiments. The set-up consisted of a nanosecond laser (CFR 400, Quantel, Paris, France; 20 Hz, 532 nm, 10 ns), UV grade collecting optic (F2, SOL instruments, Minsk, Belarus), an optical fibre (core diameter 400  $\mu\text{m}$ , Thorlabs, Newton, United States), the glass doublet ( $f = 32$  mm, Sill optics, Wendelstein, Germany), the Czerny-Turner spectrometer (Shamrock, Andor, Nottingham, United Kingdom), and a sCMOS iStar camera (Andor). A simplified scheme of LIBS experiment set-up is illustrated in Fig. 2A.

The experimental parameters used during the LIBS analysis were optimized in our previous work (Modlitbová et al., 2019) and only slight



**Fig. 3.** A1) The photograph of plant treated with UCNP dispersion (leaves – left side, root – right side). The black cross represents place with UCNP contain and the red cross represents places without UCNP contain. Following typical LIBS spectrum (black – with UCNP presence, red – without UCNP presence) with marked emission lines are at part A2. B1) The photograph of the plant treated with ion solution (leaves – left side, root – right side). The black cross represents place with Y and Yb contain and the red cross represents places without the Y and Yb contain. Following typical LIBS spectrum (black – with Y and Yb presence, red – without Y and Yb presence) with marked emission lines are at part B2. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)



**Fig. 4.** Toxicological macroscopic end-points for cabbage (*B. oleracea*) after a 72 h exposure to UCNPs dispersions and ions mixture solutions, the x axis shows nominal concentrations ( $\mu\text{g/mL}$ ), each black cross represents one sample and each red cross is a mean, the blue asterisks show notable differences (\* $p < 0.05$ , \*\* $p < 0.01$ , \*\*\* $p < 0.001$ ), A: root length (cm), B: stem length (cm), C: root + stem length (cm), D: length of the whole plant (cm). (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

modifications were made here. The 1  $\mu\text{s}$  of the gate delay, 15  $\mu\text{s}$  of the detection integration time, 20 mJ of laser pulse energy, and 20 Hz repetition rate were preserved. The energy per pulse and number of ablation layers needed for the whole plant mass removing have to be optimized again due to a different plant structure of each test plant species. The energy was set to 20 mJ per pulse which was enough for ablation of the whole plant mass in one layer. The irradiance value was approximately  $6.3 \text{ GW cm}^{-2}$ . The lateral resolution was 100  $\mu\text{m}$  based on the crater diameter. The resolution of maps is already a compromise between the laser beam energy, the size of crater diameter after the laser ablation, and the achieved sensitivity. The typical map size ranged from 150 to 400  $\text{mm}^2$  and consisted from 20,382 to 40,584 spectra; the measurements times were between 17 and 34 min.

The data from LIBS measurements of plant samples were analysed using RStudio (R-Core team, Vienna, Austria), LIBS Analyzer (CEITEC, Brno, Czech Republic), and Lightigo ImageLab (Epina, Vienna, Austria). The most appropriate emission lines were Y II 437.49 nm, Yb I 398.80 nm, and Tm I 409.42 nm. These lines were selected based on literature (Modlitbová et al., 2019). Afterwards, they were experimentally verified, i.e. 100 spectra were obtained from solid  $\text{Y}_2\text{O}_3$  (p.a., Lachema, Brno, Czech Republic),  $\text{Yb}_2\text{O}_3$  (p.a., Ubichem, Redditch, United Kingdom), and  $\text{Tm}_2\text{O}_3$  (p.a., Sigma Aldrich) standards. Spectra of selected elements were thoroughly investigated and spectral lines assigned. For those purposes we have also used Principal Component Analysis (PCA), all the spectra were processed using PCA and yielded loadings showed distinct spectral lines for individual elements (Pořízka et al., 2018). These lines were used through the whole experiment after a background subtraction using moving minimum (Yaroshchik and Eberhardt, 2014) where the size of the minimum window was set at 80 points and the smoothing window at 30 points. The data were imaged in the form of 2D maps. Each point of the map represents one obtained spectrum, thus it gives the spatial resolution. The intensity is represented by the integral of the area under the spectral line. Fig. 3 depicts the obtained spectra from plants treated with UCNPs dispersions as well as with ions mixture.

#### 2.4. Photon-upconversion microscans

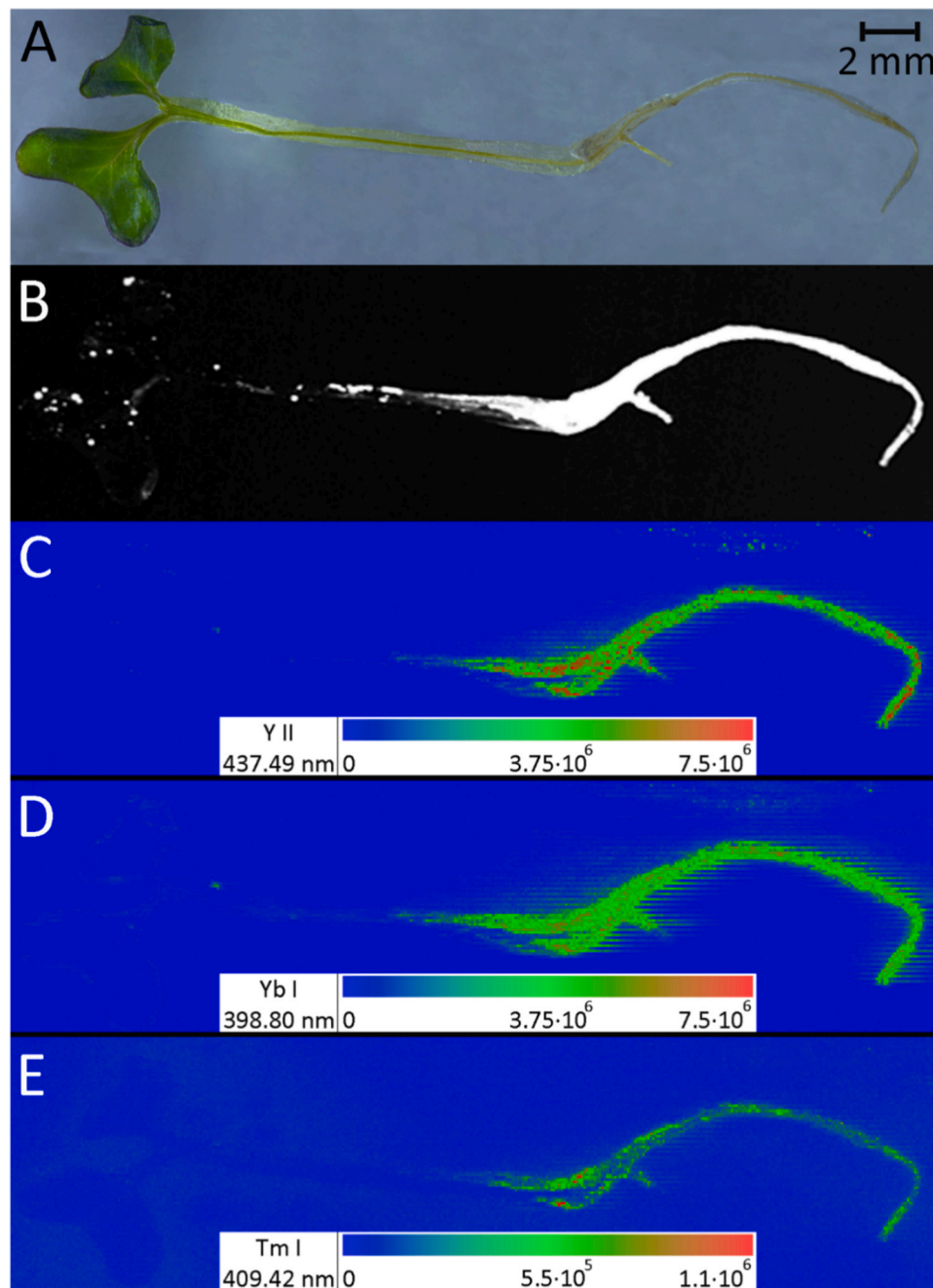
Plants exposed to UCNPs dispersions were scanned with a photon-upconversion microscanner before LIBS measurements. A laser diode (5 W, 976 nm, Besram Technology, Wuhan, China) with a focusing lens (focal length 20 mm, ThorLabs) and a 925 nm long-pass filter (Edmund Optics, Barrington, United States) was utilized as an excitation source for photon-upconversion luminescence. The detector of microscanner comprises a dichroic mirror (900 nm short pass, Thorlabs) that reflects the excitation into the microscope objective (40 $\times$ , NA 0.60). The objective collects the luminescence into a charge-coupled device array spectroscopy (QE65000, Ocean Optics, Duiven, the Netherlands) which is connected by a multimode optical fiber of 200  $\mu\text{m}$  diameter (ThorLabs), and records emission spectra in the range from 435 nm to 900 nm. The spatial resolution of the scanner was set to 40  $\mu\text{m}$  and the integration time of 10 ms was used for recording the emission spectrum in each pixel of the image. Recorded images confirmed the presence of UCNPs and evaluated the translocation of UCNPs through the root and stem to leaves. A simplified scheme of laser microscanner is shown in Fig. 2B.

### 3. Results and discussion

#### 3.1. The properties of tested compounds

The photon-upconversion maximum and the hydrodynamic diameter of UCNPs in both concentrations was established before and after the exposures to *B. oleracea* (summarized in the Supplementary data file in Table S1). No evident changes before and after the exposure were noted in photon-upconversion maximum values, and only minimal changes were registered in the hydrodynamic diameter of NPs. This demonstrated a great stability of tested UCNPs with low tendencies to agglomerate/aggregate after a contact with the plant roots. Also, the nominal concentrations of Y, Yb, and Tm ions in UCNPs dispersions and positive control solutions are listed in the Supplementary data file in Table S2.





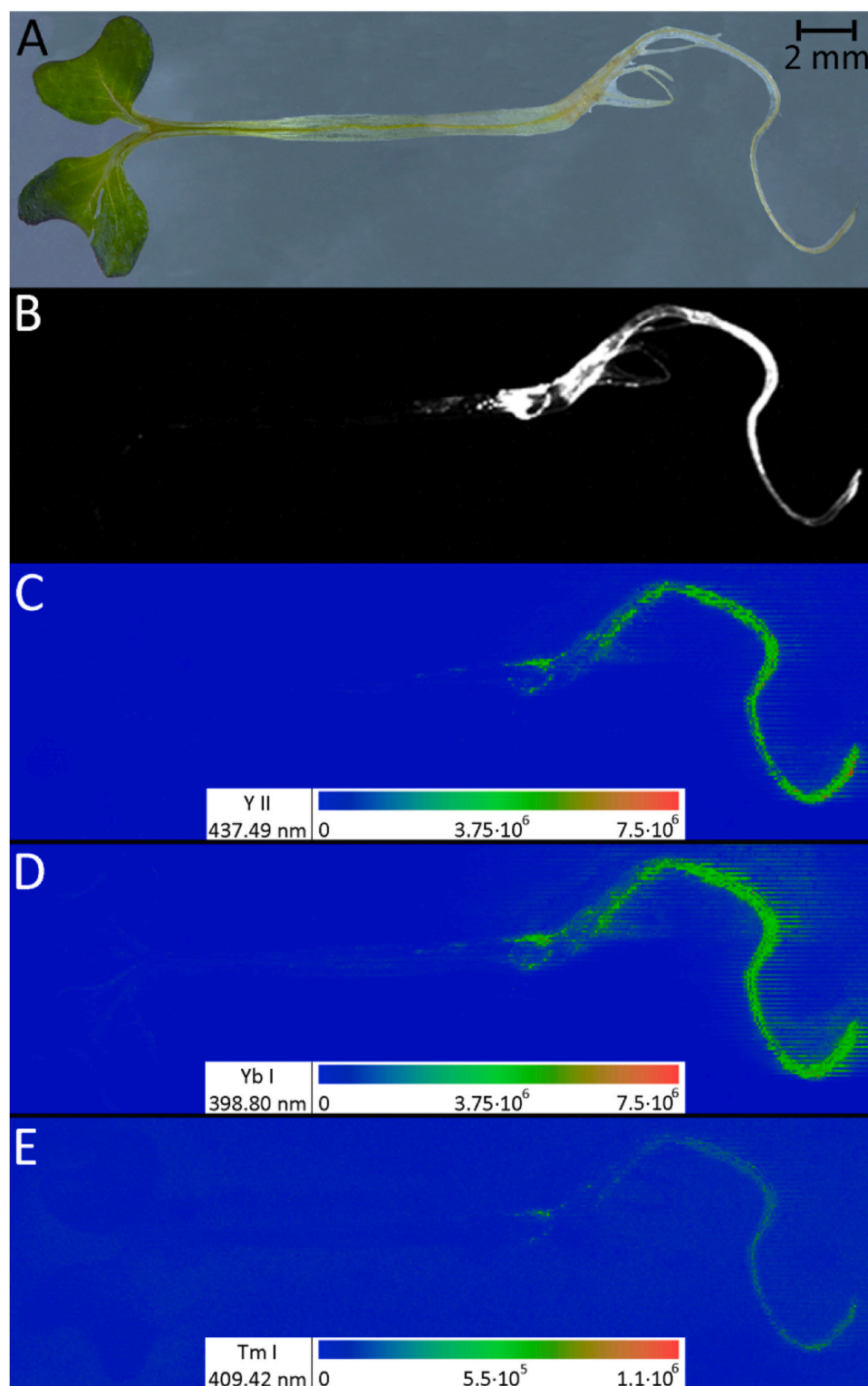
**Fig. 5.** *Brassica oleracea* after a 72-h exposure to 500 µg UCNPs/mL dispersion, A: photography of plant, B: photon-upconversion microscan of plant, C: 2D LIBS map of spatial yttrium distribution (Y II 437.49 nm), D: 2D LIBS map of spatial ytterbium distribution (Yb I 398.80 nm), E: 2D LIBS map of spatial thulium distribution (Tm I 409.42 nm). The scales show the intensity of emission lines (counts).

### 3.2. Toxicity experiments

The following parameters served as the macroscopic toxicity endpoints; the root length, stem length, root + stem length, and the whole plant length. Their significant affection was noticeable only in concentration “500” µg/mL of ions mixture solution (100 µg Y/mL + 43.8 µg Yb/mL). The root length and root + stem length (Mann-Whitney *U* test,  $^{**}p < 0.01$ ), and the stem and the whole plant length (Mann-Whitney *U* test,  $^{***}p < 0.001$ ) were significantly negatively affected (Fig. 4). On the contrary, UCNPs dispersions in both tested concentrations (500 and 50 µg UCNPs/mL) had a slightly (not statistically significant) positive effect on the growth of all plant parts. The zero effect for all monitored parameters was noticed in concentration “50” µg/mL

of ions mixture solution (10 µg Y/mL + 4.38 µg Yb/mL).

The plant growth enhancement is expected in a low UCNPs exposure dosage. Concentration 10 µg UCNPs/mL promoted the development of mung beans (Peng et al., 2012), development of soybean (Yin et al., 2015), or the radish root growth (Modlitbová et al., 2019). Higher concentrations (100, 500, or 1000 µg UCNPs/mL) showed to have an inhibitory growth effect for exposed plants. In accordance with literature, the concentration 50 µg UCNPs/mL had no negative effect on *B. oleracea* growth. On the contrary to literature, *B. oleracea* growth was not inhibited even in the concentration 500 µg UCNPs/mL. This inconsistency could be partially affected by a slightly different type of tested UCNPs; NaYF<sub>4</sub>:Yb<sup>3+</sup>,Er<sup>3+</sup>,Tm<sup>3+</sup> coated with citric acid (Peng et al., 2012), NaYF<sub>4</sub>:Yb<sup>3+</sup>,Er<sup>3+</sup> coated with polyethyleneimine (Yin et al.,

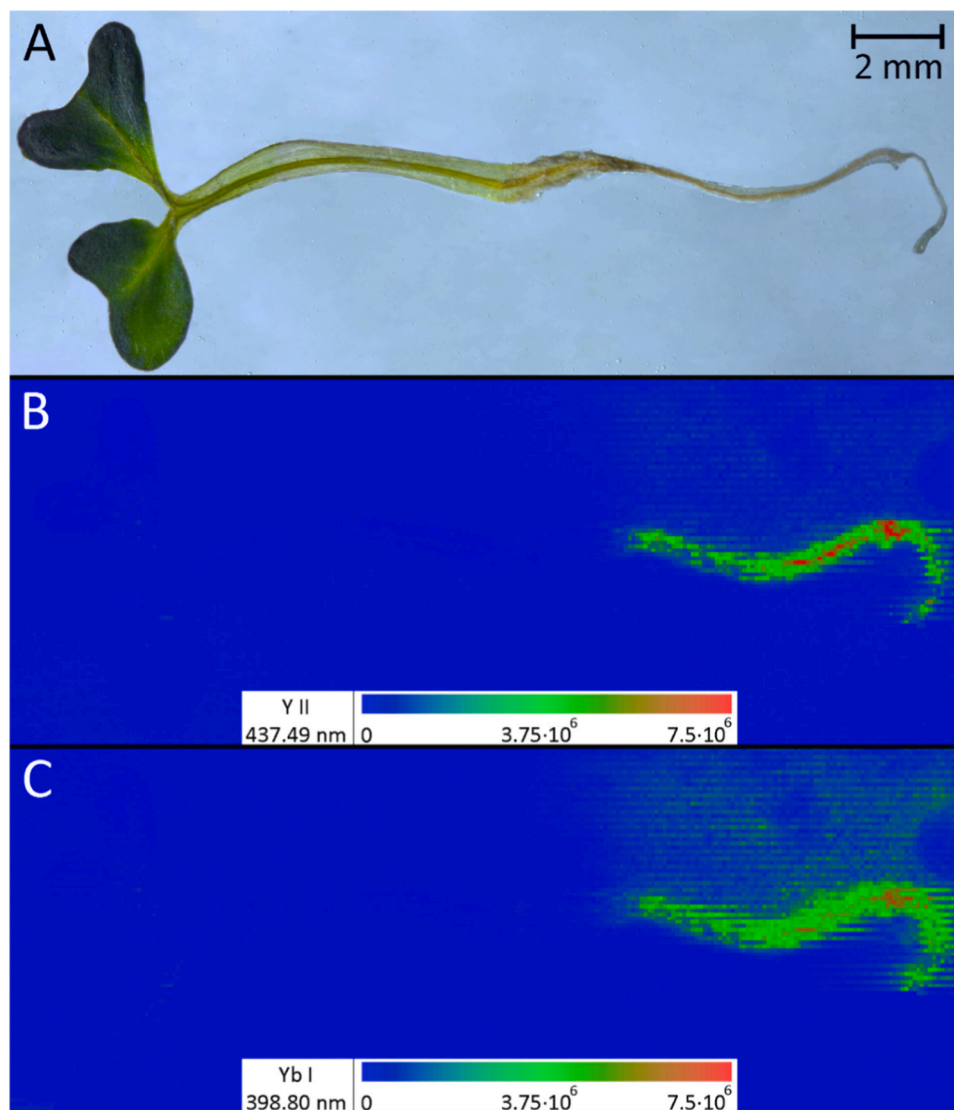


**Fig. 6.** *Brassica oleracea* after a 72-h exposure to 50 µg UCNPs/mL dispersion, A: photograph of plant, B: photon-upconversion microscan of plant, C: 2D LIBS map of spatial yttrium distribution (Y II 437.49 nm), D: 2D LIBS map of spatial ytterbium distribution (Yb I 398.80 nm), E: 2D LIBS map of spatial thulium distribution (Tm I 409.42 nm). The scales show the intensity of emission lines (counts).

2015), and  $\text{NaYF}_4:\text{Yb}^{3+},\text{Er}^{3+}$  coated with carboxylated silica shell (Modlitbová et al., 2019). Also, it could be caused by the great stability of tested UCNPs as described in previous chapter.

As it was already proved in literature (Ma et al., 2010), the effects of NPs containing rare-earth elements are not related to free ions that are released from NPs into their surroundings but by NPs themselves.

Nevertheless, these NPs have no inert shell on the surface or another type of encapsulation in order to be more stable and consequently improve their stability (Ma et al., 2010). This corresponds to previous phytotoxicity studies with UCNPs (Yin et al., 2015; Modlitbová et al., 2019), and the same behaviour is expected here due to the same silica encapsulation of UCNPs as shown previously (Modlitbová et al., 2019).



**Fig. 7.** *Brassica oleracea* after a 72-h exposure to “50” Y and Yb ions mixture solution, A: photography of plant, B: 2D LIBS map of spatial yttrium distribution (Y II 437.49 nm), C: 2D LIBS map of spatial ytterbium distribution (Yb I 398.80 nm). The scales show the intensity of emission lines (counts).

In accordance with literature, the UCNPs showed a lower toxicity than Y ions (Yin et al., 2015). This situation occurred ordinarily even for other types of NPs (Pokhrel and Dubey, 2013).

### 3.3. LIBS bioimaging

The emission signals for yttrium (Y II 437.49 nm) and for ytterbium (Yb I 398.80 nm) were accumulated separately after a proper background subtraction using moving minimum for each *B. oleracea* part (root, stem, and leaves). They were normalized to maximum value as shown in the Supplementary data file in Table S3. Then, 2D LIBS maps were constructed for selected elements that UCNPs contained, Y (Y II 437.49 nm), Yb (Yb I 398.80 nm), and Tm (Tm I 409.42 nm). These maps were completed by photon-upconversion microscans to confirm the presence of selected elements in the NPs form. The distribution of elements in *B. oleracea* exposed to 500 µg UCNPs/mL is shown in Fig. 5. The distribution of elements in *B. oleracea* exposed to 50 µg UCNPs/mL is shown in Fig. 6. Also, 2D maps of Y (Y II 437.49 nm) and Yb (Yb I 398.80 nm) were constructed for both positive controls. Fig. 7 shows Y and Yb maps for plant after exposure to “50” µg/mL ions mixture solution.

The UCNPs were transferred from the root via stem into leaves by

vascular bundles, which are the main transport corridors for water, inorganic ions, and also for UCNPs (Peng et al., 2012). However, the main bioaccumulation storage organ was the root, which is easily visible on 2D LIBS maps and followingly confirmed also by accumulated signals for Y and Yb as listed in the Table S3. The preference of bioaccumulation sites of UCNPs and ions was as follows. In lower test concentration 50 µg/mL, ions and UCNPs showed the same behaviour in plant translocation, root > stem > leaves, the detected signal from the stem and leaves was negligible. The plants exposed to the 500 µg UCNPs/mL demonstrated different trends. The order of preference was: root > stem > leaves. The same preference of storage organs was present for *R. sativus* exposed to 10 µg UCNPs/mL and to “10” µg/mL ions mixture solution (Modlitbová et al., 2019). The plants exposed to the “500” ions solution slightly differed: root > stem = leaves. The trend of bioaccumulation of ions in leaves was shown also for *R. sativus* plants exposed to the ions mixture at a high concentration 1000 µg/mL (Modlitbová et al., 2019). On the contrary, *R. sativus* exposed to 100 a 1000 µg UCNPs/ had a preference in bioaccumulation sites in this order: root > leaves > stem (Modlitbová et al., 2019). This variance might be caused by the different toxicity of UCNPs for plant growth between *R. sativus* and *B. oleracea* plants.

As visible in Figs. 5 and 6, 2D maps of spatial Y, Yb, and Tm



distribution obtained by LIBS method showed a very similar sensitivity in comparison to photon-upconversion microscans. Moreover, LIBS method could easily detect the UCNPs without any luminescence, or UCNPs whose luminescence quench. Also, LIBS offered the bioimaging of selected elements in their ionic forms as showed in the Fig. 7. Thus, LIBS is a robust method for multi-element bioimaging, which is non-dependent on the form of analysed substances.

#### 4. Conclusion

Herein, we report on the toxicity and biodistribution of UCNPs dispersed in aqueous medium for a model crop plant *B. oleracea*. The toxicity was set via four macroscopic end-points (root length, stem length, root + stem length, and whole plant length). Based on our findings, we draw several conclusions. (i) The stability of UCNPs in an aquatic dispersion during the exposure, in the direct contact with plant roots, and within test organisms was proved. (ii) The bioaccumulation of rare-earth elements in plants is concentration dependant. (iii) The UCNPs and Y and Yb ions could be transferred from roots through the stem into leaves, however the preferred storage organ is the root. (iv) Tested rare-earth elements are more toxic in ionic forms; actually, UCNPs did not show any toxicity in tested concentrations. (v) The minimal differences in preferences of Y and Yb bioaccumulation sites were observed for the plants exposed to UCNPs or to ions mixtures in the concentration 50 µg/mL. In higher tested concentration 500 µg/mL the change was visible, the UCNPs as well as ions were transferred into leaves via stem. In general, it is an extremely challenging task to compile a comprehensive view of UCNPs behaviour in major food crops. Main reasons are the complexity of the problem and the limited information available in the literature. The need of further investigation is apparent.

In this work, we presented LIBS as a fast, straightforward, and quite simple analytical method for semi-quantitative multi-element bioimaging in plant tissues. Also, LIBS enables bioimaging of various NPs after their luminescence quenching and also of NPs without any luminescence, as well as of elements in all other forms. Also, we highlight the possibility of an easy and cheap plant sample preparation with the opportunity for a long-term storage. Finally, LIBS allows the *in-situ* analysis of fresh plants without any sample preparation (Zhao et al., 2016).

#### CRedit authorship contribution statement

Pavla Modlitbová conceived and designed the experiments; Sára Strážská and Antonín Hlaváček performed the experiments; David Procházka evaluated the LIBS data; Pavla Modlitbová wrote the original draft; Antonín Hlaváček and Pavel Pořízka edited the draft; Pavel Pořízka and Jozef Kaiser supervised all steps.

#### Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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#### Appendix A. Supporting information

Supplementary data associated with this article can be found in the online version at doi:10.1016/j.ecoenv.2021.112113.

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